

## Product Information & Manual

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### Lactate Oxidase (LOX)

Cat no. LDG0033RG

#### Product Overview

##### Specification

Appearance	Yellowish amorphous powder, lyophilized
Activity	200 U/mg or more

##### Properties

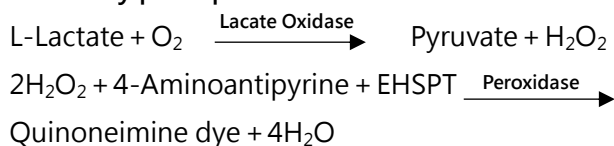
Stability	Stable at -20°C for at least one year
Molecular weight	40 kDa
Isoelectric point	5.45

##### Applications

1. Enzymatic determination of L-Lactate
2. Biosensor development <sup>(1)</sup>
3. Lactate detection in food industry <sup>(1)</sup>
4. Detection of lactate concentration in blood as a diagnostic parameter <sup>(1)</sup>

##### Assay

###### 1. Assay principle



#### 2. Unit definition

One unit causes the formation of one micromole of hydrogen peroxide (half a micromole of quinoneimine dye) per minute under the conditions described below.

#### 3. Reagent

<b>A. DL-Lactate solution</b>	0.125 M [120 mg of DL-lithium lactate (MW=96.01)/10 mL of 50 mM K-Phosphate buffer pH 7.5] (Should be prepared fresh)
<b>B. 4-AA solution</b>	0.5% (500 mg of 4-aminoantipyrine/100 mL of H <sub>2</sub> O) (Store at 4°C in a brownish bottle)
<b>C.EHSPT (TOOS) solution</b>	20 mM [296 mg N-ethyl-N-(2-hydroxy-3-sulfopropyl)-m-toluidine (MW=295.3)/50 mL of H <sub>2</sub> O] (Store at 4°C in a brownish bottle)
<b>D. Peroxidase solution</b>	25 U/mL [Prepare a stock ca. 20 mg of horseradish peroxidase (300 units/mg)/2 mL of H <sub>2</sub> O, and dilute the stock to 25 U/mL]
<b>E. SDS solution</b>	0.25% (500 mg sodium dodecyl sulfate/200 mL of H <sub>2</sub> O)
<b>F. Enzyme diluent</b>	20 mM K-Phosphate buffer, pH 7.0 containing 0.1% (w/v) sodium cholate

#### 4. Procedure

- (1) Prepare the following working solution

immediately before use and equilibrate at 37°C for approximately 5 minutes (for 8 reactions).

### Working Solution

DL-Lactate solution (Reagent A)	1.6 mL
4-AA solution (Reagent B)	0.24 mL
EHSPT solution (Reagent C)	0.16 mL
Peroxidase solution (Reagent D)	0.4 mL
Distilled water	1.6 mL
<b>Total</b>	<b>4 mL</b>

- (2) Pipette 0.5 mL of working solution into a tube.
- (3) Add 0.025 mL of the enzyme solution\* and mix with a gentle inversion.

Concentration in a reaction	
K-phosphate buffer	20 mM
DL-Lactate	48 mM
4-Aminoantipyrine	1.2 mM
EHSPT	0.76 mM
Peroxidase	2.4 U/mL

- (4) After exactly 15 minutes at 37°C, add 1 mL of SDS solution (**Reagent E**) to stop the reaction and measure the optical density at 555 nm against water (OD test).
  - (5) At the same time, prepare the blank by using the same method as the test except that the enzyme diluent (**Reagent F**) is used instead of the enzyme solution (OD blank).
- \* Dissolve the enzyme preparation in ice-cold enzyme diluent (**Reagent F**) dilute to 0.04–0.1 U/mL with the same buffer and store on ice.
- (6) Activity can be calculated by using the following formula:

**Volume activity (U/mL) =**

$$\frac{\Delta OD (OD \text{ test} - OD \text{ blank}) \times V_t \times df}{34.3 \times 1/2 \times t \times 1.0 \times V_s}$$

$$= \Delta OD \times 0.237 \times df$$

**Weight activity (U/ mg) =** (U/mL) × 1/C

V<sub>t</sub>: Total volume (1.525 mL)

V<sub>s</sub>: Sample volume (0.025 mL)

34.3: Millimolar extinction coefficient of quinoneimine dye under the assay condition (cm<sup>2</sup>/micromole)

1/2: The factor is derived from the stoichiometric relationship in which one mole of H<sub>2</sub>O<sub>2</sub> yields half a mole of quinoneimine dye.

t: Reaction time (15 minutes)

1.0: Light path length (cm)

df: Dilution factor

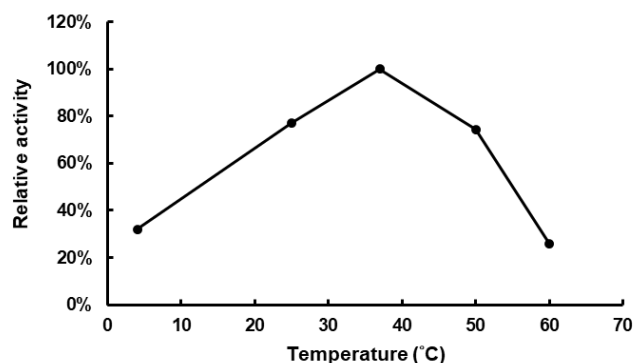
C: Enzyme concentration in dissolution (mg/mL)

### Reference

1. *Agustina Godino, et al.* His-tagged lactate oxidase production for industrial applications using fed-batch fermentation. *Journal of Biotechnology* (2023).

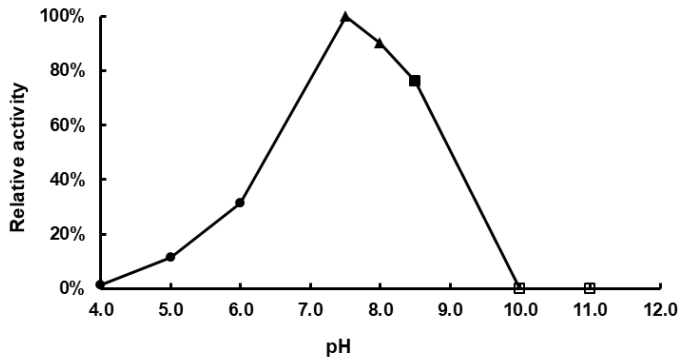
### The effect of different conditions on Lactate Oxidase

A.



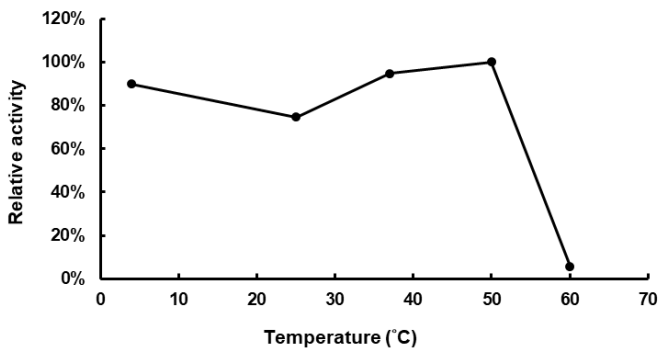
**Figure A. Temperature activity of LOX.** The enzyme reactions in 20 M K-Phosphate buffer, pH 7.5, were carried out under different temperatures.

B.



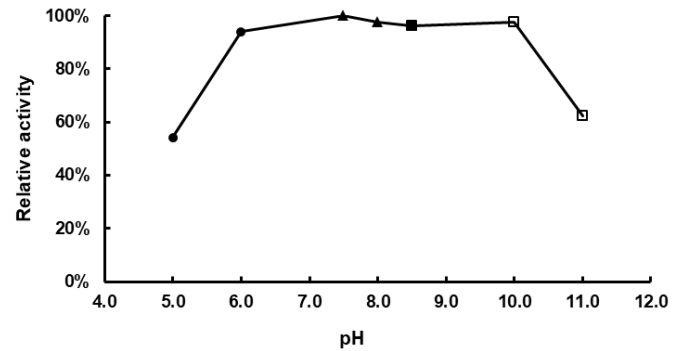
**Figure B. pH activity of LOX.** The buffer conditions with various pH values were used in the reaction at 37°C. pH 4.0-6.0, 0.1 M Sodium citrate buffer; pH 7.5-8.0, 0.1 M Potassium phosphate buffer; pH 8.5, 0.1 M Tris-HCl buffer; pH 10.0-11.0, 0.1 M Carbonate-bicarbonate buffer.

C.



**Figure C. Thermal stability of LOX.** The enzyme powder was reconstituted by double-distilled water and treated with different temperatures for 10 minutes. Final concentration: 10 U/mL

D.



**Figure D. pH stability of LOX.** The enzyme powder was reconstituted by double-distilled water and treated with different pH buffer condition at 25°C for 16 hours. pH 4.0-6.0, 0.1 M Sodium citrate buffer; pH 7.5-8.0, 0.1 M Potassium phosphate buffer; pH 8.5, 0.1 M Tris-HCl buffer; pH 10.0-11.0, 0.1 M Carbonate-bicarbonate buffer.

#### Disclaimer

For Research Use or Further Manufacturing Only.

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